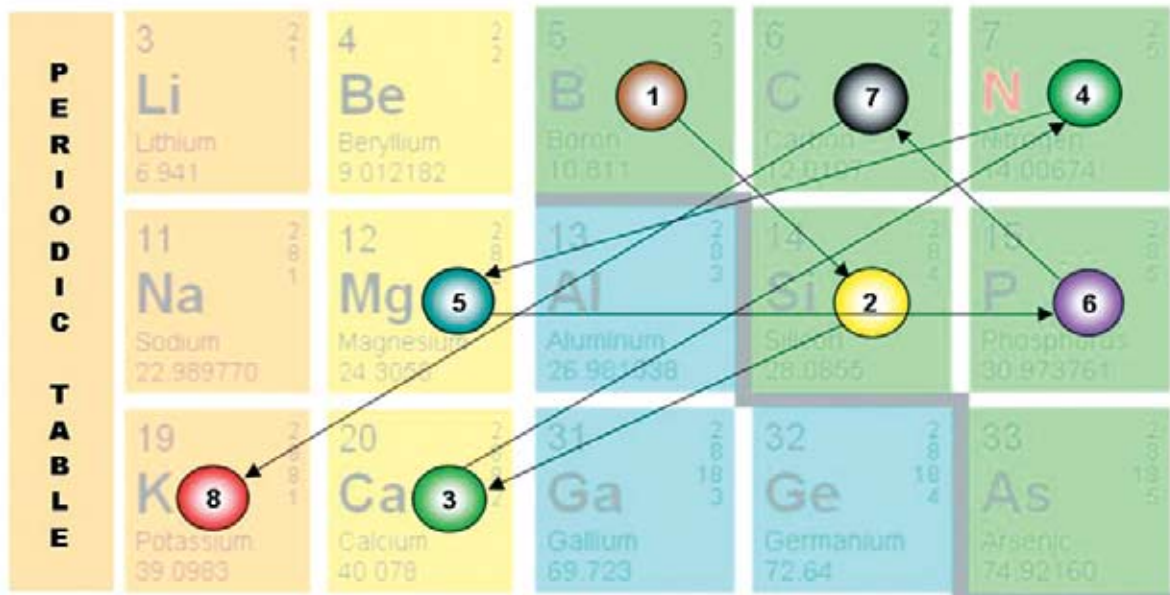


Biochemical sequence of nutrition in plants



Plant biochemical sequences begin with:

1. **Boron**, which activates
2. **Silicon**, which carries all other nutrients, starting with
3. **Calcium**, which binds
4. **Nitrogen** to form amino acids, DNA and cell division.

Amino acids form proteins such as chlorophyll and tag trace elements, especially

5. **Magnesium**, which transfers energy via
6. **Phosphorus** to
7. **Carbon** to form sugars which to where
8. **Potassium** carries them. This is the basis of plant growth.

The biochemical sequence

By HUGH LEVEL

IN FEBRUARY of 1994 at the Austin, Texas Eco Fair, I lunched with Neal Kinsey, one of America's top soil consultants. Neal was lecturing about the key importance of calcium in the early stages of fruit development where cell division occurs. His metaphor was that an apple not much bigger than a prune had virtually all the calcium it would get by harvest. He tested soils for calcium and applied it as needed, but unfortunately this did *not* guarantee that sufficient calcium ended up in the apple.

When I asked him what he did in regard to

boron, which was responsible for sap pressure, he responded, "Of course, boron is necessary for calcium uptake, and we test for boron. If it is needed we put it there, but we still can't guarantee that calcium gets in the apple."

Hmmm. So I asked what he did about silicon. My biodynamic experience showed silicon was the basis of transport in both plants and animals. Neal's response was classic, "We don't test for silicon. It's in all soils, whether sand or clay."

Until then it hadn't sunk home with me that I was used to looking for the visual signs of silicon in plants and I hadn't actually seen any soil or leaf tests that included it. This got me wondering, and

as I investigated I found, almost uniformly, soil and leaf testing labs did not test for silicon unless it was specifically requested.

As a biodynamic grower I was annoyed. Biodynamic forerunner Rudolf Steiner, with a doctorate in maths, chemistry and biology, identified the oxides of calcium and silicon, lime and silica, as the opposite poles of life chemistry. I'd used this concept for years and years, along with Jochen Bockemühl's leaf studies from his book *In Partnership with Nature* and Johann W von Goethe's treatise *The Metamorphosis of Plants* as guides. Neal's comment that he didn't test for silicon caught me by surprise. But, on the other hand, my university curriculum was biochemistry rather than agricultural chemistry so I hadn't realized what 19th Century agricultural chemistry taught. Looking further, I found that in the early days of agricultural chemistry, Justus von Liebig tested both soils and plants for silicon, found it in all cases, was unable to prove it was an essential nutrient by excluding it from plant media and thereafter dropped it from his tests. This became the norm for agricultural testing.

Neal Kinsey, with his riddle of getting calcium into early fruit development, got me thinking.

Gradually I realised there was an obvious hierarchy of how elements worked in living organisms. One thing had to occur before the next thing could happen, and on down the line in a sequence. In 2004 I put together a PowerPoint slide show for Graeme Sait's agronomy team at Nutri-Tech in Yandina, Queensland, and in it I summarised this hierarchy of elements, calling it the biochemical sequence.

I told the Nutri-Tech agronomists that *boron* kicks off this sequence by activating silicon, making it an amorphous fluid and providing sap pressure. I knew that boron was used in making glass, which is amorphous fluid silica, and I'd found this relationship also held true for plant chemistry.

Of course, sap pressure would be no use without a transport system to contain it, and *silicon* enables the actual transport of nutrients. Interestingly, applying too much boron too early in a crop cycle is notable for burning seedlings and young transplants – such as sprouting squash, beans or tomatoes – because too much sap pressure in such a tiny plant drives sodium out to the leaf margins. Nevertheless, in plants where leaf veins are highly branched (such as flowering beans, squash and

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tomatoes) boron is important in later growth to maintain strong enough sap pressure to make such a complex system work.

On the other hand, highly siliceous plants, such as grasses, need less boron to give them sap pressure since their transport vessels all run parallel without branching. That's like irrigation lines that only feed one sprinkler head: it doesn't take much pressure. An exception is bananas, which have a huge transport system with lots of fluid flow. They need plenty of boron to send calcium and amino acids all the way to the top of the bell stalk for cell division to occur in the bunch.

Obviously, without robust transport nowhere near as much nutrient reaches the leaves or is stored in the fruits. Chemical agriculture gets around this to some extent since, even with a weak transport system, anything that is highly soluble (such as potassium nitrate) is simply taken up along with water. Though this dilutes the sap, it flows quite easily due to low sap density. This is why (synthetic) chemically grown foods commonly have a coarse, watery cell structure as well as lower nutrition and poorer keeping quality. However, without a robust transport system,

heavier nutrients such as calcium, magnesium, complex carbohydrates and amino acids can easily be left behind.

Third in the biochemical sequence is *calcium*. This is the last thing you want to leave behind because of its role in nitrogen fixation and amino acid chemistry. Calcium balances charge in proteins and is particularly important in cell division, which is the first thing that happens in fruit or seed formation after pollination. Without it there would be no fruit or seed.

For example, in maize calcium leaf test targets are between 0.3% and 1%, increasing as the maize approaches tasselling with the higher target range more desirable during kernel formation. If calcium does not reach the ear in sufficient quantities, the kernels near the end of the ear simply do not fill out. With a crop like soybeans, double or even triple the calcium values of maize are needed for full pod set without shedding pods (a common problem in soybeans). Wouldn't you like to see every kernel on your maize fill out to the end of the ear and every soybean blossom produce a pod of beans? This only happens when boron, silicon and calcium work together optimally.

As just mentioned, wherever calcium goes there

"I realised there was an obvious hierarchy of how elements worked in living organisms. One thing had to occur before the next thing could happen, and on down the line in a sequence."

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Biodynamics

also goes *nitrogen*, which is the basis of amino acid formation, protein chemistry and DNA replication. Once nitrogen enters the picture all sorts of proteins, enzymes and hormones are produced and very complex things are set in motion involving trace elements such as iron, zinc, copper, manganese, cobalt, molybdenum and so on.

Above all, there must be energy harvest or plants will never grow. Though all parts of a plant's protein chemistry require amino acid nitrogen, large amounts of amino acids go into the formation of chlorophyll where energy is gathered. Since photosynthesis requires *magnesium*, it is fifth in the biochemical sequence, ahead of all the more minor trace elements. Of course, photosynthesis is not simply a matter of chlorophyll catching energy. The energy has to be transferred into producing sugars out of carbon dioxide and water, which requires *phosphorus* for energy transfer. Otherwise the chlorophyll burns up and the leaves turn a wine-red colour.

However, as long as there is enough phosphorus, *carbon* is pried loose from carbon dioxide so it can combine with water to make sugar and release oxygen. Then the sugars pass into the plant's sap, where *potassium*, the electrolyte, conducts them to wherever they most need to go.

Understandably, this sequence is oversimplified. For example, sulphur is the classic catalyst in carbon (organic) chemistry. Without it, nothing – not even the boron – would work. Also, potassium has a very close relationship with silicon, so when silicon carries calcium and amino acids to the cell division sites in the plant, potassium plays the role of an electronic doorway that lets the calcium and amino acids enter the cells that are preparing to divide. If cold weather slows potassium down or if it is in short supply, then calcium and amino acids cannot reach the cell nuclei, the DNA cannot divide, cell division fails and the fruit falls off the plant. Sometimes entire fruit crops are lost to a couple of degrees of frost when a light spray of kelp with potassium silicate would save the day.

However, the most important thing to understand is the role of boron, silicon and calcium in the hierarchy of plant chemistry. Growers who simply feed plants nitrogen, phosphorus and potassium (NPK) tend to shortcircuit the biological processes where strong sap pressure (boron) leads to good nutrient transport (silicon), following which optimal cell division and photosynthesis occur (calcium, nitrogen, magnesium and phosphorus). Then, with high plant energy (carbon and potassium) plants are able to shed enough of their sap as root exudates to feed abundant microbial mineral release, nitrogen fixation and protozoal digestion around crop roots. This enables crops to enjoy



An ear of sweet corn with optimum boron, silicon, calcium and amino acid nitrogen.

rich nutrition and be truly healthy. This only works where boron, silicon, calcium and amino acid nitrogen (from steady microbial fixation and digestion) are all high. If calcium and amino acids are watered down with nitrate and potassium salts, sap pressure is impaired, cell division is hampered, photosynthesis is weakened, magnesium and phosphorus are diluted and we're returned to where NPK growers are today.

Comprehensive testing (the subject of another article) reveals that without taking the biochemical sequence into account it is common for plants – even in organic situations where soluble nitrogen and potassium are high – to luxury feed on nitrogen and potassium to the exclusion of calcium, magnesium, phosphorus and trace elements, particularly zinc and molybdenum. In summary, if we fail to solve Neal Kinsey's riddle, we can be caught in this situation and suffer from the conventional NPK growers' problems of pests, diseases, poor flavour and poor keeping quality. 🍌

Hugh Lovel – formally educated in maths, physics, chemistry, biology and psychology – was a biodynamic farmer for more than 30 years in the mountains of North Georgia, US, before becoming an Australian. He lives in Tolga on the Atherton Tablelands in Queensland and lectures, writes and consults. His AgPhysics advisory service covers all aspects of agriculture, including livestock and environmental repair, with special emphasis on the underlying patterns of energy that determine such things as weather, crop vigour, flavour and keeping quality.

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